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HSC Maharashtra Board question paper: March 2013

Note:

- All questions are compulsory.
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

CHEMISTRY: SECTION - I

Q.1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In body centred cubic structure the space occupied is about
(A) 68% (B) 53%
(C) 38% (D) 32%
- For a gaseous reaction the unit of rate of reaction is
(A) $L\text{ atm}^{-1}\text{ s}^{-1}$ (B) $\text{atm min}^{-1}\text{ s}^{-1}$
(C) atm s^{-1} (D) $\text{atm}^{-1}\text{ s}^{-1}$
- Which of the following compounds contains S-S as well as S-O bonds?
(A) Sulphuric acid (B) Thiophosphoric acid
(C) Sulphurous acid (D) Thiophosphorous acid
- Which of the following solutions shows maximum depression in freezing point?
(A) 0.5 M $\text{Fe}_2(\text{SO}_4)_3$ (B) 1.0 M NaCl
(C) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (D) 0.5 M BaCl_2
- For a chemical reaction, $\Delta S^\circ = 0.035\text{ kJ/K}$ and $\Delta H^\circ = -20\text{ kJ}$.
(A) 5.14 K (B) 57.14 K
(C) 571.4 K (D) 5714.0 K
- The standard e.m.f. of the following cell is 0.463 V
 $\text{Cu}|\text{Cu}^{2+}||\text{Ag}^+|\text{Ag}$, $E^\circ_{\text{Ag}^+/\text{Ag}} = 0.800\text{ V}$.
What is the standard potential of Cu electrode?
(A) 1.337 V (B) 0.337 V
(C) 0.463 V (D) -0.463 V
- Fe_2O_3 is reduced to spongy iron near the top of blast furnace by
(A) H_2 (B) CaO
(C) CO (D) CO_2

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